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metals consist of closely packed cations and loosely held valence electrons rather than neutral atoms explain briefly why metals are good conductors of electric currents metals are good conductors because electrons can flow freely in the metal name the three crystal arrangements of closely packed metal atoms

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what we assess conveys strong implicit messages to students about what matters in given course 58 63 a good portion of the students in any college level chemistry class are enrolled in order to fulfill a requirement for their major and or post baccalaureate aims not because they find the subject matter fascinating

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chapter 17 review reaction kinetics mixed review short answer answer the following questions in the space provided 1 the reaction for the decomposition of hydrogen peroxide is  $2\text{H}_2\text{O}_2(\text{aq}) \rightarrow 2\text{H}_2\text{O}(\text{l}) + \text{O}_2(\text{g})$  list three ways to speed up the rate of decomposition for each one briefly explain why it is effective based on collision theory

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these assessment tools provide teachers with feedback about how students are mastering or not mastering the concepts of chemistry a credible assessment of an overall chemistry program should be based on information from a variety of assessment tools over a span of several years

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9729 chemistry gce advanced level h2 syllabus 2021 14 for elements in group 2 magnesium to barium and group 17 chlorine to iodine candidates should be able to f describe and deduce from e values the relative reactivity of elements of i group 2 as reducing agents ii group 17 as oxidising agents

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impact of chemistry instructors and faculty on their students learning and to increase our collective understanding of how students learn chemistry and how assessment can be employed as a strategy to make substantive improvements in instruction and outcomes in chapter 2 ted clark discusses important innovations taking place in an

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1 j cal 239 1 cal j 4 18 the amount needed to increase the temp of an object exactly 1 c depends both on its mass and chemical composition heat capacity the amount of heat it takes to raise the temp of 1g of a substance 1 c specific heat

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the syllabus for h3 chemistry builds on that for h2 chemistry and includes the whole of the h2 chemistry syllabus only content that is not already part of the h2 chemistry syllabus is specifically set out here the h3 chemistry syllabus introduces additional content in two areas namely spectroscopic techniques and further organic mechanisms

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for elements in group 1 lithium to caesium and group 17 chlorine to iodine candidates should be able to describe and explain the relative reactivity of elements of group 1 as reducing agents in terms of ease of loss of electrons group 17 as oxidising agents in terms of ease of gain of electrons

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abstract pesticide additives pas are auxiliary ingredients added to the pesticide manufacturing and use processes constituting 1 to 99 of the pesticide and often composed of benzene and chlori

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