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web igsce chemistry answers b the mass of the empirical formula is 4 12 8 1 16 72 since this is equal to the relative formula mass the molecular formula is the same as the empirical formula that is c<sub>4</sub>h<sub>8</sub>o<sub>10</sub> you know the mass of anhydrous sodium sulfate 1 42 g you can work out the mass of water of crystallisation

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web 11 a 133 cs b 127 i c 31 p 3 d 57 co 3 13 a carbon 12 12 c b this atom contains six protons and six neutrons there are six electrons in a neutral 12 c atom the net charge of such  
a neutral atom is zero and the mass number is 12 c the preceding answers are correct

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0 25 2 assume x 0 25 then 1 8 10 4 x 0 25

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