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z atomic number of protons n of neutrons a z n a typical isotopic symbol takes this form ex the isotopic symbol for fluorine would be fill in the missing items in the table below fill in the missing items in the table below fill in the missing items in the table below chemistry atomic structure practice i 17 isotopic symbol na

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the atomic number tells you the number of in one atom of an element it also tells you the number of in a neutral atom of that element the atomic number gives the identity of an element as well as its location on the periodic table no two different elements will have the atomic number

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atomic structure worksheet label the parts of an atom on the diagram below 4 what type of charge does a proton have 5 what type of charge does a neutron have 6 what type of charge does an electron have 7 which two subatomic particles are located in the nucleus of an atom 8

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atomic structure worksheet draw an atom below and label the following electron proton neutron nucleus list the 3 subatomic particles their charges masses and where in the atom they are located in the chart below describe how to determine the number of each subatomic particle in an atom below

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the atomic number  $z$  tells you the number of where do you find the atomic number on the periodic table what subatomic particle differs in isotopes of a particular element how is the atomic mass the left hand superscript in a symbol determined from the structure of an atom

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basic atomic structure worksheet word bank the number of protons in one atom of an element determines the atom's and the number of electrons determines the

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3 the atomic number tells you the number of in one atom of an element it also tells you the number of in a neutral atom of that element the atomic number gives the identity of an element as well as its location on the periodic table no two different elements will

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1 draw five protons in the nucleus of the atom label them with their charge 2 draw six neutrons in the nucleus of the atom 3 draw two electrons in the first energy level and label them with their charge 4 draw three electrons in the second energy level and label them with their charge

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atomic structure name date b sketch an atom draw 5 protons in the nucleus and label with the charge draw 6 neutrons in the nucleus and label with the charge draw 2 electrons in the 1st energy level and label with their charge draw 3 electrons in the 2nd ndenergy level

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bismuth hydrogen magnesium mercury 10 name the element which has the following numbers of particles a b c d f 26  
electrons 29 neutrons 26 protons 53 proton 74 neutrons 2 electrons neutral atoms 20 protons 86 electrons 125 neutrons  
82 protons o neutrons 11 if you know only the following information can you always determine what the

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an atom is made up of protons electrons and neutrons the number of electrons will determine how atoms interact with one another and decide if the atom as a whole is positive negative or neutral the mass number of an atom is found by adding together the number of protons and neutrons

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test your knowledge of the three particles that make up an atom protons neutrons and electrons choose the correct answer from four options and get instant feedback

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