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web igsce chemistry answers b the mass of the empirical formula is 4 12 8 1 16 72 since this is equal to the relative formula mass the molecular formula is the same as the empirical formula that is $\text{C}_4\text{H}_8\text{O}_{10}$ you know the mass of anhydrous sodium sulfate 142 g you can work out the mass of water of crystallisation

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web 11 a 133 cs b 127 i c 31 p 3 d 57 co 3 13 a carbon 12 12 c b this atom contains six protons and six neutrons there are six electrons in a neutral 12 c atom the net charge of such a neutral atom is zero and the mass number is 12 c the preceding answers are correct

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