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an ion selective electrode ise also known as a specific ion electrode sie is a transducer or sensor that converts the change in the concentration of a specific ion dissolved in a solution into an electrical potential the voltage is theoretically dependent on the logarithm of the ionic activity according to the nernst equation

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as indicated by their name ion selective electrodes possess a high degree of selectivity the selectivity of the ise is determined by the composition of the membrane ideally the membrane allows the uptake of only one specific ion into it the analyte ion may be a cation or an anion

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ion selective electrode ise is an analytical technique used to determine the activity of ions in aqueous solution by measuring the electrical potential ise has many advantages compared to other techniques including it is relatively inexpensive and easy to operate it has wide concentration measurement range

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ion selective electrodes ises are potentiometric sensors that include a selective membrane to minimize matrix interferences the most common ise is the ph electrode which contains a thin glass membrane that responds to the h concentration in a solution

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ion selective electrodes ises are a modern analytical tool for the fast and direct ion activity determination of sample solutions together with a suitable ion meter or titrator ises form a well affordable and versatile measuring system

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as the name implies ion selective electrodes are devices that are selective for a particular ionic species the most well known and highly used ion selective electrode is a ph electrode which is sensitive toward the h ion a desirable aspect of ion selective electrodes is their ease of use

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