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sodium hydroxide wikipedia

Mar 26 2024

sodium hydroxide also known as lye and caustic soda is an inorganic compound with the formula naoh it is a white solid ionic compound consisting of sodium cations na and hydroxide anions oh sodium hydroxide is a highly corrosive base and alkali that decomposes lipids and proteins at ambient temperatures and may cause severe chemical

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Feb 25 2024

sodium hydroxide naoh a corrosive white crystalline solid that contains the na cation and the oh anion it readily absorbs moisture until it dissolves sodium hydroxide is the most widely used industrial alkali and is often used in drain and oven cleaners

sodium hydroxide chemical safety facts

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updated october 14 2022 sodium hydroxide sodium hydroxide naoh also known as caustic soda or lye is a highly versatile substance used in a variety of manufacturing processes sodium hydroxide is a co product of chlorine production key points overview uses benefits safety information faqs related the american vision modern

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sodium hydroxide is a white translucent crystalline solid it is commonly referred to as caustic soda due to its corrosive action on many substances it decomposes proteins at room temperatures and may cause chemical burns to human bodies

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sodium hydroxide is very corrosive it is generally used as a solid or a 50 solution other common names include caustic soda and lye sodium hydroxide is used to manufacture soaps rayon paper explosives dyestuffs and petroleum products

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close sodium hydroxide is a highly corrosive base and alkali that decomposes lipids and proteins at ambient temperatures and may cause severe chemical burns it is highly soluble in water and readily

absorbs moisture and carbon dioxide from the air it forms a series of hydrates naoh nh2o

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sodium hydroxide naoh also known as lye or caustic soda is an extremely caustic corrosive and damaging to human tissue white solid that readily dissolves in water sodium hydroxide is used in the manufacture of soaps rayon and paper in petroleum refining and in homes as drain cleaners and oven cleaners sodium hydroxide is one of the

sodium hydroxide formula properties application

Jul 18 2023

chemical properties sodium hydroxide is a strong base and ranks as one of the most commonly used alkalis it has a high affinity for water undergoing exothermic reactions when combined with it which can be used to generate considerable amounts of heat this is often exploited in many applications that require the increase of temperature

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sodium hydroxide is an alkali which is also known as caustic soda caustic means burning and caustic soda takes its name from the way it can burn the skin it has the chemical formula of naoh it is a base meaning it has a high ph it is the most common thing used to raise the ph of solutions for example to neutralize an acid

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sodium hydroxide naoh exhibits hygroscopic properties leading to its ability to readily absorb moisture from the surrounding atmosphere the substance shows a high degree of corrosiveness thereby posing the potential to induce significant skin burns and irritation to the ocular and other anatomical regions

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chemical properties sodium hydroxide is completely ionic containing sodium ions and hydroxide ions the hydroxide ion makes sodium hydroxide a strong base which reacts with acids to form water and the corresponding salts for example with hydrochloric acid sodium chloride is formed naoh aq hcl aq nacl aq h 2 o l

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sodium hydroxide is overall considered safe for general use but only in small amounts at lower concentrations at higher concentrations sodium hydroxide is unsafe it can cause chemical

hydroxide wikipedia

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sodium hydroxide is a multi million ton per annum commodity chemical the corresponding electrically neutral compound ho is the hydroxyl radical the corresponding covalently bound group oh of atoms is the hydroxy group both the hydroxide ion and hydroxy group are nucleophiles and can act as catalysts in organic chemistry

2023-07-21

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ph scale acids bases ph and buffers article khan academy

Jan 12 2023

to give you the short answer an acidic solution has a high concentration of hydrogen ions h greater than that of pure water a basic solution has a low h concentration less than that of pure water to see where this definition comes from let s look at the acid base properties of water itself autoionization of water

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sodium hydroxide is used to manufacture soaps rayon paper explosives dyestuffs and petroleum products it is also used in processing cotton fabric laundering and bleaching metal cleaning and processing oxide coating electroplating and electrolytic extracting it is commonly present in commercial drain and oven cleaners top of page

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Nov 10 2022

overview cas no 1310 73 2 sodium hydroxide naoh has no smell it is made of solid white crystals that absorb water from the air sodium hydroxide is caustic sodium hydroxide can harm workers who

come in contact with it the level of harm depends upon the amount duration and activity

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Oct 09 2022

chemical element human nutrition sodium deficiency sodium 22 sodium 24 sodium na chemical element of the alkali metal group group 1 ia of the periodic table sodium is a very soft silvery white metal sodium is the most common alkali metal and the sixth most abundant element on earth comprising 2 8 percent of earth s crust

4 strong and weak bases chemistry libretexts

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4 strong and weak bases page id jim clark truro school in cornwall this page explains the terms strong and weak as applied to bases as a part of this it defines and explains k b and pk b the usual way of comparing the strengths of bases is to see how readily they produce hydroxide ions in solution

hydroxide alkaline ph base britannica

Aug 07 2022

sodium hydroxide naoh also known as caustic soda or lye is of great industrial importance calcium barium and strontium all alkaline earth metals form soluble hydroxides that are strong bases but are less stable than the alkali hydroxides of these calcium hydroxide ca oh 2 commonly known as slaked lime is the most common

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