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review of basic chemistry answer key name the following elements spelling counts write the symbols for the following elements hydrogen sulphur cl chlorine carbon nitrogen na sodium pb lead ag silver sodium phosphorus fluorine magnesium potassium calcium zinc iron 3

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the following are examples of official ap chemistry questions in multiple choice short response and long response format i ll go over the answers in detail to give you a sense of the types of problems you ll face on the test and how you might solve them

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1 a solution can vary in composition while a compound cannot vary in composition solutions are homogeneous at the molecular level while other mixtures are heterogeneous 3 a the process is endothermic as the solution is consuming heat

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1 0m what is the percent by mass of a water solution that contains 60 g of calcium chloride CaCl_2 and that has a mass of 400g 15 what is the percent by mass of a solution containing 25 g of NaCl in 175 g of solution 14 you need to prepare 500 0 ml of 0.750M NaBr solution using a stock solution with a concentration of 2.50M NaBr

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base your answers to questions 16 through 18 on the information below in an investigation a dripless wax candle is massed and then lighted as the candle burns a small amount of

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the a level h2 chemistry examination consists of four papers three theory papers and one practical assessment paper 1 is a 1 hour long paper containing 30 compulsory multiple choice questions carrying a total of 30 marks there will be five to eight multiple completion type questions

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here are some things about the exams that you should know the h1 chemistry exam has two papers paper one has a weightage of 33 and consists of thirty mcqs paper 2 has a weightage of 67 and it consists of open ended and structured questions h2 chemistry has four papers paper 1 has 15 weightage and 30 mcqs

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the syllabus for h3 chemistry builds on that for h2 chemistry and includes the whole of the h2 chemistry syllabus only content that is not already part of the h2 chemistry syllabus is specifically set out here the h3 chemistry syllabus introduces additional content in two areas namely spectroscopic techniques and further organic mechanisms

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while physics and chemistry were a bit lengthy and moderate to difficult biology was easy compared to the other two subjects out of the three sections chemistry was the toughest read detailed

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