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7 18a calculate the mass of carbon in the sample of carbon dioxide using the same formula and your answers to the previous two calculations $\text{g C} = \frac{27.3}{100} \times 13 \text{ g CO}_2 = 0.273 \times 13 \text{ g} = 0.035 \text{ g C}$ calculate the percentage of carbon that was in the original sample of sodium bicarbonate using the following formula

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