# Free pdf Chapter 10 chemical quantities test b (Read Only)

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express your answer to the correct number of significant figures one milligram is equal to 0 001 grams drag each number to the correct location sort the numbers based on the number of significant figures they have

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3 a 1 20 x 10 24 atoms of fe b 2 63 x 10 24 atoms of si c 7 23 x 10 21 atoms of li d 5 27 x 10 23 s 2 ions 4 a 5 72 x 10 22 molecules of water h 2 o b 1 93 x 10 21 formula units of sodium chloride nacl c 1 24 x 10 24 formula units of calcium bromide cabr 2 d 8 43 x 10 22 molecules of glucose c 6 h 12 o

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chemical equations also provide us with the relative number of particles and moles that react to form products in this section you will explore the quantitative relationships that exist between the quantities of reactants and products in a balanced equation this is known as stoichiometry

#### chapter 7 chemical reactions and chemical quantities Dec 26 2023

this chapter will describe how to symbolize chemical reactions using chemical equations how to classify some common chemical reactions by identifying patterns of reactivity and how to determine the quantitative relations between the amounts of substances involved in chemical reactions that is the reaction stoichiometry

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topics on chapter 10 test chemical quantities 10 1 the mole I o 10 1 1 convert among the count mass and volume of something I o 10 1 2 explain how chemists count the number of atoms molecules or formula units in a substance I o 10 1 3 determine the molar mass of an element and of a compound 10 2 mole mass and mole volume relationships

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note since scientists often measure the mass of substances in grams not moles it is typical to be given the starting mass of a substance in grams and must first convert to moles before using the mole ratio ex given the balanced equation below 2h2 o2 2h2o

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although there are untold millions of possible chemical reactions most can be classified into a small number of general reaction types classifying reactions has two purposes it helps us to recognize similarities among them and it enables us to predict the products of certain reactions

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a chemical reaction is described by a chemical equation an expression that gives the identities and quantities of the substances involved in a reaction a chemical equation shows the starting compound s the reactants on the left and the final compound s the products on the right

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