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the specific heat of the human body calculated from the measured values of individual tissues is 2 98 kj kg 1 c 1 this is 17 lower than the earlier wider used one based on non measured values of 3 47 kj kg 1 c 1 the contribution of the muscle to the specific heat of the body is approximately 47 and the contribution

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the si unit of specific heat capacity is joule per kelvin per kilogram j kg 1 k 1 for example the heat required to raise the temperature of 1 kg of water by 1 k is 4184 joules so the specific heat capacity of water is 4184 j kg 1 k 1

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